EFFECTS OF ASPHALT CEMENT REJUVENATING AGENTS

FINAL REPORT

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Pax International, Anaheim, California
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METRIC CONVERSION FACTORS*

To Convert from	<u>To</u>	Multiply by	
	<u>Length</u>		
foot inch yard mile (statute)	<pre>meter (m) millimeter (mm) meter (m) kilometer (km)</pre>	0.3048 25.4 0.9144 1.609	
	Area		
square foot square inch square yard	square meter (m²) square centimeter (cm²) square meter (m²)	0.0929 6.451 0.8361	
	Volume (Capacity)		
<pre>cubic foot gallon (U.S. liquid)** gallon (Can. liquid)** ounce (U.S. liquid)</pre>	cubic meter (m³) cubic meter (m³) cubic meter (m³) cubic centimeter (cm³)	0.02832 0.003785 0.004546 29.57	
	Mass		
ounce-mass (avdp) pound-mass (avdp) ton (metric) ton (short, 2000 1bm)	gram (g) kilogram (kg) kilogram (kg) kilogram (kg)	28.35 0.4536 1000 907.2	
	Mass per Volume		
<pre>pound-mass/cubic foot pound-mass/cubic yard pound-mass/gallon (U.S.)** pound-mass/gallon (Can.)**</pre>	kilogram/cubic meter (kg/m³) kilogram/cubic meter (kg/m³) kilogram/cubic meter (kg/m³) kilogram/cubic meter (kg/m³)	16.02 0.5933 119.8 99.78	
<u>Temperature</u>			
deg Celsius (C) deg Fahrenheit (F) deg Fahrenheit (F)	kelvin (K) kelvin (K) deg Celsius (C)	t _k =(t _c +273.15) t _k =(t _F +459.67)/1.8 t _c =(t _F -32)/1.8	

^{*}The reference source for information on SI units and more exact conversion factors is "Metric Practice Guide" ASTM E 380.

 $[\]star\star$ One U.S. gallon equals 0.8327 Canadian gallon.

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ABSTRACT

Louisiana's initial work in the recycling of asphaltic concrete pavements has demonstrated the need to obtain a base of knowledge in the area of rejuvenating age-hardened reclaimed asphalt cement. In this report, eight rejuvenating agents are examined with regard to the rejuvenator's effect, at various levels of addition, upon age-hardened asphalts; the subsequent aging characteristics of the rejuvenated asphalts; and the uniformity of mixing of the rejuvenating agents with asphaltic concrete mixes. Prediction equations yielding the proper quantity of rejuvenator to be added to an oxidized asphalt were examined. The rejuvenated asphalts followed anticipated patterns when tested for penetration at 77°F, viscosity at 140°F and 275°F, and ductility at 77°F. Generally, the rejuvenated asphalts demonstrated acceptable viscosity indices when subjected to the Thin Film Oven Test. Testing for uniformity of mixing in asphaltic concrete mixes proved inconclusive.

IMPLEMENTATION STATEMENT

The recommendations of this report call for the Department to utilize the results of this study in the development of the Department's recycling program. A direct application of the prediction equation confirmed in this report will provide the necessary information for recycle mix design. The viscosity indices examined will provide a means to approve rejuvenating agents for a qualified products list.

INTRODUCTION

The present-day awareness of the need to conserve available raw materials and energy, along with the current economics of highway construction, has evolved recycling of asphaltic concrete pavements. Innovations in milling equipment for the removal of existing pavements efficiently and economically have created an accessible supply of materials for these recycling efforts. Most of the recovered paving mixtures, however, have an asphalt binder that has hardened due to aging. A logical step to be taken when reworking the mixture is to add a softening or rejuvenating agent to increase the penetration and lower the viscosity of the aged binder. Such rejuvenation should not only increase the workability of the recycled mix at the time of reconstruction, but should also yield added service life to the resulting asphalt binder.

This study is concerned with acquiring a basic familiarity with several asphalt cement rejuvenating agents in anticipation of their future use in Louisiana recycling operations. Of primary importance is the development of a consistent relationship between aged asphalt cement and a rejuvenating agent which would provide an approximation of when and in what quantities these additives may be incorporated in recycled mixtures. The effect of rejuvenating agents on the subsequent aging characteristics of the binder and the uniformity of mixing of the rejuvenating agents in asphaltic concrete mixes is examined.

SCOPE

This laboratory study was conducted in three phases:

Phase I consisted of the determination of the rejuvenating agents effect, at various levels of addition, upon age-hardened asphalts from different sources. Eight rejuvenators, each at four different levels of addition, were combined with each of two asphalt cements which had been laboratory aged to two different states of oxidation. The combined binder was then characterized by viscosities at 140°F and 275°F, penetration at 77°F, and ductility at 77°F.

Phase II investigated the subsequent aging characteristic of the rejuvenated asphalts. The Thin Film Oven Test was used to determine the aging index of the two original asphalts and of the rejuvenated asphalts.

Phase III included the testing for uniformity of mixing of the rejuvenating agents with asphaltic concrete mixes. Marshall test properties were used for this evaluation.

The intent of this study was not to evaluate one manufacturer against another, but rather to obtain basic knowledge for subsequent decision making.

METHODOLOGY

A request was made to various companies in regard to the use of their rejuvenating agents under the auspices of this study and the scope presented herein. In response, ten rejuvenators from six manufacturers were forwarded. Two of these rejuvenators were eliminated from consideration: the first was an emulsion and thus deemed inappropriate as a softening agent for a hot plant recycling process; the second had a low flash point (below 350°F).

Sufficient supplies of two AC-30 grade asphalt cements—Exxon and Lion—were obtained. Originally, it was thought that these asphalts could be aged in a prolonged thin film oven exposure. The logistics of aging enough asphalt for the study negated this idea. It was then decided to place an amount of the Exxon asphalt in a shallow aggregate sample pan and subject this binder to continuous 400°F oven heating. The material was agitated twice daily to prevent a crust from forming on the surface and to ensure that the material in the pan was being oxidized uniformly. Each morning a sample was taken and the viscosity at 140°F was tested. The following schedule was obtained:

Viscosity (Poise)
6,525
13,328
24,245
44,881
92,132

Based on this data, it was assumed that the viscosity would approximately double during each twenty-four-hour period. As a representative sampling of oxidized pavement materials to be recycled might typically yield viscosities of 50,000 and 100,000 poises, it was decided to choose aging periods of 96 and 120 hours.

Phase I

The asphalt cements from the two different sources were aged to two different states of oxidation according to the above plan. These oxidized states were classified by viscosity at 140°F and 275°F, penetration at 77°F, and ductility at 77°F (Table 1).* Original viscosities of each of the rejuvenating agents were then determined (Table 2). At that point, it was necessary to choose the four levels of rejuvenator addition.

Previous work published by various authors** has examined the relationship between a blended oxidized asphalt cement and a rejuvenating agent. This relationship has been defined in the form of three equations which can be used to predict the rejuvenator content necessary to produce a desired viscosity:

$$\log(V) = a + bp \tag{a}$$

$$\log - \log(V) = a + bp$$
 (b)

$$log-log(V) = a + b(log p)$$
 (c)

where:

V = viscosity of the blend (measured at 140°F in centipoises)

p = volume percent rejuvenator in blend

a and b = constants to be determined for each asphaltrejuvenator-blend

^{*}All tables may be found in Appendix A (page 13).

^{**&}quot;Evaluation of Selected Recycling Modifiers," Holmgreen, R. J., and Epps, J. A., prepared for National Cooperative Highway Research Program.

As equation (b) was more familiar to the authors, it was used to determine the preliminary relationship between the rejuvenators obtained for this study and each of the two oxidized states of the two source asphalts. Axes were plotted on semi-log paper with the viscosity axis ranging from 10¹ to 10⁸ centipoises and the percent rejuvenator axis ranging from 0 to 100 percent. One end point was determined from the viscosity of the oxidized asphalt (0 percent rejuvenator); the other end point was determined from the viscosity of the rejuvenator (100 percent rejuvenator). An example of this procedure is shown in Figure 1.* Based on the relationship determined in this manner, the four levels of addition were chosen for each rejuvenator. One level for each rejuvenator was selected to bring the oxidized asphalt back to its original (pre-oxidized) state. The other levels were chosen to be both reasonable (with respect to quantity) and to provide a good spread along the curve. Table 3 gives the four addition levels selected for further testing for each of the rejuvenators.

Quantities of the oxidized asphalts were combined with the selected addition levels of each rejuvenator and were placed in separate containers. The matrix for testing purposes for each rejuvenator is represented by that for Rejuvenator A (Figure 2). Each of the blends was classified by testing for viscosity at 140°F and 275°F, penetration at 77°F, and ductility at 77°F.

Phase II

The Thin Film Oven Test was used to determine the aging index of each of the original asphalt cements. This index is simply the oven-aged viscosity divided by the original viscosity and is intended to classify asphalts by characteristic hardening susceptibilities.

^{*}All figures may be found in Appendix B (page 29).

The corresponding aging indices for the rejuvenated asphalts were also determined. For each rejuvenator, the resulting rejuvenated asphalt whose viscosity most closely matched that of the original asphalt was chosen for testing. The index was found by dividing the viscosity of the thin film oven-aged rejuvenated asphalt by the viscosity of the rejuvenated asphalt.

Phase III

In this phase, Marshall test properties were determined and compared for the following identical mixes, with the same mixing and compaction temperature being used for all mixes:

- (1) briquettes made with new asphalt;
- (2) briquettes made with a rejuvenated binder; the binder was composed of oxidized asphalt, pre-blended with the appropriate amount of rejuvenator to obtain the same viscosity as the new asphalt (as determined by the method described in Phase I); and
- (3) briquettes made with a rejuvenated binder; in this case, the appropriate amount of rejuvenator was added to a mixture of oxidized asphalt and aggregate.

This matrix was examined five times with three different rejuvenating agents being blended with the first oxidation level of Exxon AC-30 and two different rejuvenators being blended with the second oxidation level of Lion AC-30. The five rejuvenators were chosen to represent the full viscosity range of the rejuvenators indicated in Table 2 (page 16).

DISCUSSION OF RESULTS

Phase I

Tables 4 through 7 (pages 18-21) present the results determined from the classification testing accomplished on the rejuvenated asphalts. The results show the anticipated increase in penetration and decrease in viscosity with greater levels of rejuvenator concentra-Ductilities were not run on the lowest and highest levels of concentration; these levels were chosen to obtain a good spread of viscosity data for curve fitting and would not be utilized as viable blends in hot mix design. It should be noted that the viscosities associated with blends Fl, Gl and Hl in Table 7 (251032, 245537 and 292929 poises, respectively) exhibit values greater than the oxidized asphalt cement itself (195313 poises from Table 1). absolute viscosities were tested according to ASTM D 2171, which states its applicability to materials having viscosities less than 200,000 poises. It is possible that the accuracy in testing near this limit may be such that the value of the viscosity of the oxidized asphalt cement was actually outside the test limit, thereby allowing the inconsistently high values observed in Table 7.

The viscosity at $140^{\circ}F$ generated in this phase was used to examine the relationship between viscosity and percent rejuvenator addition as described in the Methodology section. Each prediction equation was examined to determine which equation would more closely approximate the data. The original asphalt cement and rejuvenator viscosity were plotted on axes correlated to equation (a) and equation (b) (page 4). Viscosity data was plotted on the corresponding graphs. Examples are presented in Figures 3 and 4 (pages 33 and 34). Equation (c), $\log - \log(V) = a + b(\log p)$, was not considered as the $\log (p)$ is not defined when no rejuvenator is considered (p = 0).

Due to the different axis scales and the difficulty of graphically determining whether a given data point was closer to the prediction curve from equation (a) or (b), the difference between actual and predicted viscosities from each equation was numerically examined. A smaller absolute value of this difference was considered indicative of the better relationship. Table 8 (page 22) presents a step-by-step example of this procedure. The data generated by this analysis is shown in Table 9. Ninety percent of the blends examined in this phase demonstrated equation (b) to be better able to predict the observed viscosity.

Phase II

The viscosity indices (V.I.) for the original Exxon and Lion asphalt cements were 1.80 and 1.90, respectively. Figures 5 through 8 (pages 35-38) present the viscosity indices of the rejuvenated asphalts for each state of artificial oxidation. The solid bar represents the viscosity of that rejuvenated asphalt which most closely matches, for each rejuvenator, the viscosity of the original asphalt cement. The clear portion of each bar shows the viscosity of the rejuvenated asphalt after thin film oven testing.

A viscosity index of 4.0 has been established as the maximum allowable for use in Louisiana. Rejuvenator H exceeds this limit in Figure 8 and approaches this limit in the other three cases. While rejuvenators F and G do not exceed the maximum allowable index, they do approach the limit especially for the highly oxidized original asphalts (second state of oxidation). The viscosity indices for the remaining rejuvenated asphalts were acceptable, and these rejuvenators would be expected to perform similarly to new asphalt cements.

Phase III

Marshall briquettes were formed according to the design indicated in the methodology and were tested for stability and flow. Mean Marshall properties are presented in Table 10. As shown, the

briquettes made with the pre-blended binder have slightly higher stabilities than those briquettes where the rejuvenator was added to the mixture of aggregate and oxidized asphalt. Although it is possible that the rejuvenating agents, in the case where they were added separately to the mix, had not fully blended with the oxidized asphalt at the time of testing, it is felt that the slight difference in stabilities between the pre-blended binder and the separately added rejuvenator can be attributed to normal testing error.

No discernible differences in mix properties were observed between the tested rejuvenator types.

CONCLUSIONS

The following conclusions are drawn from the data generated in this study and, as such, are confined to the source and grade of asphalt cement examined.

- 1. The prediction equation, log-log(V) = a + bp, can be used as a first approximation for the determination of rejuvenating agent requirements.
- 2. The addition of an appropriate quantity of rejuvenating agent to an oxidized asphalt can provide a binder whose properties closely match those of an original asphalt.
- 3. All of the rejuvenating agents with the exception of rejuvenator H displayed acceptable hardening characteristics when subjected to the Thin Film Oven Test.
- 4. Testing for uniformity of mixing of the rejuvenating agents with asphaltic concrete mixes was inconclusive.

RECOMMENDATIONS

The properties of the aged asphalt cement in the reclaimed hot mix will, for the majority of cases, be in need of some type of rejuvenation in order to be considered as an effective binder in a recycled mix. With this in mind and recognizing the conclusions drawn from this study, the following recommendations are made:

- 1. As shown in this study, a standardized viscosity agehardening index (V.I.) should be utilized for the qualification of rejuvenating agents.
- 2. The log-log(V) = a + bp prediction equation should be used as a first approximation to determine the quantity of rejuvenating agent necessary in recycled mixes.
- 3. The uniformity of rejuvenator mixing with the oxidized asphalt cement in reclaimed hot mix should be examined further. It is recommended that samples be taken and analyzed from a recycling project in which the rejuvenating agent is introduced to the mix by different methods.

APPENDIX A

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TABLE 1
ORIGINAL AND OXIDIZED ASPHALT CEMENT PROPERTIES

Asphalt Cement	Oxidation State	Penetration (77°F)	Viscosity (Poise @ 140°F)	Viscosity (Saybolt Furol Seconds @ 275°F)	Ductility (77°F)
Exxon	New	60	3,151	302	100+
Exxon	lst	19	57,872	1,134	15
Exxon	2nd	10	125,555	1,626	.წ
Lion	New	54	2,673	252	100+
Lion	lst	15	64,736	1,017	10
Lion	2nd	10	195,313	*	4

^{*}This sample too hard to test.

TABLE 2
REJUVENATOR VISCOSITIES

Rejuvenator	Viscosity (Poise @ 140°F)
A	1.67
В	110.67
С	1.27
D	25.02
E	85.0
F	0.19
G	0.30
H	0.12

TABLE 3
REJUVENATOR ADDITION LEVELS

Rejuvenator /Level	Percent <u>Addition</u>	Rejuvenator /Level	Percent Addition
Al	5	El	15
A2	13	E2	30
A3	21	E3	45
A4	29	E4	60
В1	20	Fl	3
B2	35	F2	9
В3	50	F3	15
B4	65	F4	21
C1	5	G1	3
C2	13	G2	9
C3	21	G3	15
C4	29	G4	21
Dl	15	Hl	3
D2	25	H2	9
D3	35	Н3	15
D4	45	H4	21

TABLE 4

PROPERTIES OF REJUVENATED EXXON FIRST OXIDATION STATE

Rejuvenated Asphalt	Penetration (77°F)	Viscosity (Poise @ 140°F)	Viscosity (Saybolt Furol Seconds @ 275°F)	Ductility (77°F)
A1 A2 A3 A4	27 53 119 238	17,447 4,390 1,236 433	655 342 188 110	100+ 100+ -
B1 B2 B3 B4	31 51 86 150	12,094 4,239 1,533 602	522 325 205 135	100+ 100+ -
C1 C2 C3 C4	26 58 119 247	21,527 4,162 1,175 398	709 327 176 102	100+ 100+
D1 D2 D3 D4	35 67 108 190	8,846 2,941 1,258 533	449 260 165 108	100+ 100+ -
E1 E2 E3 E4	31 56 104 184	12,686 3,575 1,171 494	518 266 148 94	100+ 100+ -
F1 F2 F3 F4	25 52 118 258	24,436 5,462 1,460 418	795 405 221 129	100+ 100+ -
G1 G2 G3 G4	22 44 98 180	29,186 6,685 1,654 672	829 423 217 137	100+ 100+ -
H1 H2 H3 H4	27 60 142 266+*	26,607 4,276 1,056 304	842 373 198 114	100+ 100+ -

^{*}The penetration needle hit the bottom of the sample cup.

TABLE 5

PROPERTIES OF REJUVENATED EXXON - SECOND OXIDATION STATE

Rejuvenated Asphalt	Penetration (77°F)	Viscosity (Poise @ 140°F)	Viscosity (Saybolt Furol Seconds @ 275°F)	Ductility (77°F)
A1 A2 A3 A4	19 41 79 192	39,916 7,547 2,092 573	957 455 231 135	100+ 100+ -
B1 B2 B3 B4	23 38 73 127	18,489 7,185 1,900 720	710 393 235 149	100+ 100+
C1 C2 C3 C4	17 44 95 217	35,194 7,026 1,739 492	910 431 218 115	100+ 100+ -
D1 D2 D3 D4	27 48 98 227	13,466 4,821 1,590 660	589 334 198 126	- 100+ 100+ -
E1 E2 E3 E4	24 45 84 170	18,384 4,870 1,608 490	700 330 178 98	- 100+ 100+ -
F1 F2 F3 F4	16 45 98 218	41,675 7,817 1,917 556	1,074 467 234 135	100+ 100+ -
G1 G2 G3 G4	16 39 79 167	58,433 10,386 2,730 825	1,131 527 290 155	100+ 100+
H1 H2 H3 H4	16 46 99 245	69,133 7,162 1,967 452	1,249 475 260 129	100+ 100+ -

TABLE 6

PROPERTIES OF REJUVENATED LION FIRST OXIDATION STATE

Rejuvenated Asphalt	Penetration (77°F)	Viscosity (Poise @ 140°F)	Viscosity (Saybolt Furol Seconds @ 275°F)	Ductility (77°F)
A1 A2 A3 A4	24 46 100 206	21,880 4,957 1,343 444	619 312 170 104	100+ 100+ -
B1 B2 B3 B4	30 45 77 137	14,771 5,000 1,689 384	508 308 195 129	100+ 100+ -
C1 C2 C3 C4	22 53 115 244	20,661 4,448 1,086 364	605 291 151 88	100+ 100+
D1 D2 D3 D4	33 50 95 185	10,636 3,491 1,382 511	421 361 246 103	100+ 100+
E1 E2 E3 E4	30 51 113 187	11,382 3,568 1,030 464	432 240 130 84	100+ 100+
F1 F2 F3 F4	23 55 114 255+*	31,563 4,907 1,256 341	727 333 180 98	100+ 100+ -
G1 G2 G3 G4	21 46 94 195	27,398 5,648 1,585 519	670 338 191 113	- 100+ 100+ -
H1 H2 H3 H4	25 58 125 255+*	26,491 3,778 1,080 280	680 290 166 87	100+ 100+

^{*}The penetration needle hit the bottom of the sample cup.

TABLE 7

PROPERTIES OF REJUVENATED LION SECOND OXIDATION STATE

Rejuvenated Asphalt	Penetration (77°F)	Viscosity (Poise @ 140°F)	Viscosity (Saybolt Furol Seconds @ 275°F)	Ductility (77°F)
A1 A2 A3 A4	15 29 56 118	143,760 17,306 3,916 1,013	1,446 569 278 150	100+ 100+ -
B1 B2 B3 B4	18 31 52 105	58,752 11,487 3,041 850	916 478 260 155	100+ 100+ -
C1 C2 C3 C4	15 27 63 135	128,552 19,628 3,388 904	1,390 572 254 136	100+ 100+ -
D1 D2 D3 D4	20 31 58 107	55,274 13,443 3,520 1,136	886 455 242 139	100+ 100+ -
E1 E2 E3 E4	19 32 65 139	76,117 11,318 2,481 710	1,033 422 195 106	100+ 100+ -
F1 F2 F3 F4	15 30 52 117	251,032 29,430 6,352 1,464	* 851 399 193	- 40 100+ -
G1 G2 G3 G4	13 24 48 88	245,537 38,371 5,966 1,872	* 906 362 211	- 48 100+ -
H1 H2 H3 H4	14 27 62 148	292,929 22,584 3,748 914	* 948 303 159	90 100+

^{*}These samples were too hard to test.

TABLE 8

PREDICTION EQUATION ANALYSIS

Observed Viscosity (V_0) = 423,900 cp @ p = 35 Oxidized Asphalt Viscosity $(V_A) = 5,787,200$ cp @ p = 0 Rejuvenator Viscosity (V_R) = 11,067 cp @ p = 100

Determination of Coefficients

Equation (a)

$$\log(V_A) = a + b(0) = 6.76$$

$$a = 6.76$$

$$log(V_R) = 6.76 + b(100) = 4.04$$
 $log-log(V_R) = .830 + b(100)$

$$b = -.027$$

Equation (b)

$$log(V_A) = a + b(0) = 6.76$$
 $log-log(V_A) = a + b(0) = .830$

$$a = .830$$

$$log - log(V_p) = .830 + b(100)$$

$$b = -.0022$$

Prediction of Viscosity (V_p) @ p = 35

Equation (a)

$$V_{\rm p} = 6.53 \times 10^5 \text{ cp}$$

Equation (b)

$$log(V_p) = 6.76 + (-.027)(35)$$
 $log-log(V_p) = .830 + (-.0022)(35)$

$$V_p = 4.6 \times 10^5 \text{ cp}$$

Difference Between Observed and Predicted Viscosities

Equation (a)

$$|V_0 - V_p| = 2.29 \times 10^5 \text{ cp}$$

$$= 2,290$$
 poise

Equation (b)

$$|V_0 - V_p| = 3.57 \times 10^4 \text{ cp}$$

= 357 poise

TABLE 9

COMPARISON OF OBSERVED AND PREDICTED VISCOSITIES*

a. REJUVENATED EXXON - FIRST OXIDATION STATE

Rejuve- nator	Addition Level (%)	Viscosity Observed (V _o)	Viscosity Eq. (a) (V _a)	Viscosity Eq. (b) (V _b)	V _o -V _a	v _o -v _b
A1	5	17,446	34,300	25,000	16,800	7,520
A2	13	4,390	15,000	7,180	10,600	2,790
A3	21	1,236	6,530	2,290	5,300	1,060
A4	29	433	2,850	807	2,420	374
B1	20	12,094	16,600	12,900	4,500	771
B2	35	4,239	6,530	4,600	2,290	357
B3	50	1,533	2,570	1,770	1,040	237
B4	65	602	1,010	731	408	129
C1	5	21,527	33,500	23,700	12,000	2,200
C2	13	4,162	14,100	6,360	9,940	2,200
C3	21	1,175	5,930	1,920	4,750	745
C4	29	398	2,490	645	2,100	247
D1	15	8,846	17,800	12,500	8,940	3,610
D2	25	2,941	8,130	4,880	5,190	1,940
D3	35	1,258	3,720	2,040	2,460	778
D4	45	533	1,700	901	1,170	368
E1	15	12,686	21,900	16,700	9,190	4,030
E2	30	3,575	8,320	5,350	4,740	1,770
E3	45	1,171	3,160	1,870	1,990	701
E4	60	494	1,200	713	708	219
F1	3	24,436	39,400	27,100	14,900	2,650
F2	9	5,462	18,400	6,660	12,900	1,200
F3	15	1,460	8,610	1,870	7,150	412
F4	21	418	4,030	593	3,610	175
G1	3	29,186	39,900	28,800	10,700	378
G2	9	6,685	19,200	7,880	12,500	1,200
G3	15	1,654	9,230	2,410	7,570	760
G4	21	672	4,440	820	3,760	148
H1	3	26,607	38,800	25,000	12,200	1,640
H2	9	4,276	17,700	5,350	13,400	1,070
H3	15	1,056	8,040	1,340	6,980	289
H4	21	304	3,660	391	3,350	87

^{*}All viscosities are presented in poises. All viscosities have been rounded to three significant digits with the exception of the observed viscosity.

TABLE 9 (CONTINUED)

COMPARISON OF OBSERVED AND PREDICTED VISCOSITIES

b. REJUVENATED EXXON - SECOND OXIDATION STATE

Rejuve- nator	Addition Level (%)	Viscosity Observed (V _O)	Viscosity Eq. (a) (Va)	Viscosity Eq. (b) (V _b)	v _o -v _a	v _o -v _b
A1	5	39,916	71,600	50,000	31,700	10,100
A2	13	7,547	29,000	12,900	21,500	5,320
A3	21	2,092	11,800	3,730	9,710	1,640
A4	29	573	4,780	1,210	4,200	634
B1	20	18,489	30,200	22,600	11,700	4,070
B2	35	7,185	10,400	7,050	3,170	140
B3	50	1,900	3,550	2,410	1,650	510
B4	65	720	1,220	901	496	181
C1	5	35,194	70,800	47,400	35,600	12,200
C2	13	7,026	28,200	11,300	21,200	4,320
C3	21	1,739	11,200	3,100	9,480	1,360
C4	29	492	4,470	957	3,970	465
D1	15	13,466	35,100	22,600	21,600	9,090
D2	25	4,821	15,000	7,980	10,100	3,160
D3	35	1,590	6,380	3,040	4,790	1,450
D4	45	660	2,720	1,240	2,060	580
E1	15	18,384	41,700	30,600	23,300	12,300
E2	30	4,870	13,800	8,500	8,930	3,630
E3	45	1,608	4,570	2,630	2,960	1,020
E4	60	490	1,510	901	1,020	411
F1	3	41,675	84,300	55,200	42,700	13,600
F2	9	7,817	37,800	12,200	30,000	4,400
F3	15	1,917	17,000	3,130	15,100	1,210
F4	21	556	7,620	915	7,060	359
G1	3	58,433	85,500	58,900	27,100	475
G2	9	10,386	39,400	14,600	29,100	4,170
G3	15	2,730	18,200	4,080	15,500	1,350
G4	21	825	8,390	1,280	7,570	456
H1	3	69,133	83,200	50,700	14,000	18,400
H2	9	7,162	36,300	9,700	29,100	2,540
H3	15	1,967	15,800	2,220	13,900	249
H4	21	452	6,920	593	6,470	141

TABLE 9 (CONTINUED)

COMPARISON OF OBSERVED AND PREDICTED VISCOSITIES

c. REJUVENATED LION - FIRST OXIDATION STATE

Rejuve- nator	Addition Level (%)	Viscosity Observed (V _o)	Viscosity Eq. (a) (Va)	Viscosity Eq. (b) (V _b)	V _o -V _a	v _o -v _b
A1	5	21,800	38,000	27,200	16,100	5,300
A2	13	4,957	16,300	7,570	11,300	2,610
A3	21	1,343	6,980	2,350	5,640	1,010
A4	29	444	2,990	809	2,550	365
B1	20	14,771	17,800	13,300	3,010	1,480
B2	35	5,000	6,760	4,530	1,760	472
B3	50	1,689	2,570	1,670	881	19
B4	65	384	977	668	593	284
C1	5	20,661	37,600	26,300	16,900	5,610
C2	13	4,448	15,800	6,980	11,400	2,530
C3	21	1,086	6,650	2,090	5,570	1,000
C4	29	364	2,800	696	2,430	332
D1	15	10,636	20,000	13,700	9,320	3,090
D2	25	3,491	9,120	5,350	5,630	1,850
D3	35	1,382	4,170	2,220	2,790	834
D4	45	511	1,910	975	1,390	464
E1	15	11,382	23,700	18,500	12,300	7,070
E2	30	3,568	8,710	5,860	5,140	2,290
E3	45	1,030	3,200	2,040	2,170	1,010
E4	60	464	1,170	770	711	306
F1	3	31,563	44,200	29,700	12,600	1,850
F2	9	4,907	20,700	7,110	15,700	2,200
F3	15	1,256	9,660	1,950	8,400	696
F4	21	341	4,520	607	4,180	266
G1	3	27,398	44,800	31,900	17,400	4,540
G2	9	5,648	21,500	8,660	15,900	3,010
G3	15	1,585	10,400	2,630	8,770	1,050
G4	21	519	4,980	886	4,460	367
H1	3	26,491	43,600	27,600	17,100	1,160
H2	9	3,778	19,800	5,860	16,000	2,080
H3	15	1,080	9,020	1,460	7,940	379
H4	21	280	4,100	421	3,820	141

TABLE 9 (CONTINUED)

COMPARISON OF OBSERVED AND PREDICTED VISCOSITIES

d. REJUVENATED LION - SECOND OXIDATION STATE

Rejuve- nator	Addition Level (%)	Viscosity Observed (V _O)	Viscosity Eq. (a) (Va)	Viscosity Eq. (b) (V _b)	V _o -V _a	v _o -v _b
A1	5	143,760	180,000	74,200	35,400	69,500
A2	13	17,306	42,400	17,500	25,100	196
A3	21	3,916	16,600	4,710	12,700	792
A4	29	1,013	6,470	1,430	5,460	415
B1	20	58,752	42,700	29,600	16,100	29,100
B2	35	11,487	13,600	8,230	2,160	3,250
B3	50	3,041	4,370	2,560	1,320	485
B4	65	850	1,400	877	546	27
C1	5	128,552	107,000	71,600	21,400	57,000
C2	13	19,628	41,100	16,100	21,500	3,560
C3	21	3,388	15,800	4,150	12,400	764
C4	29	1,136	6,050	1,220	4,920	84
D1	15	55,274	50,700	32,300	4,570	23,000
D2	25	13,443	20,700	10,800	7,210	2,670
D3	35	3,520	8,410	3,900	4,890	381
D4	45	1,136	3,430	1,520	2,290	385
E1	15	76,117	60,300	44,200	15,900	32,000
E2	30	11,318	18,600	11,300	7,300	18
E3	45	2,481	5,750	3,270	3,270	788
E4	60	710	1,780	1,060	1,070	345
F1	3	251,032	129,000	83,500	122,000	168,000
F2	9	29,430	56,200	17,000	26,800	12,400
F3	15	6,352	24,500	4,080	18,200	2,270
F4	21	1,464	10,700	1,130	9,250	339
G1	3	245,537	131,000	90,200	115,000	155,000
G2	9	38,371	58,600	21,000	20,200	17,300
G3	15	5,966	26,300	5,590	20,300	371
G4	21	1,872	11,800	1,680	9,930	193
H1	3	292,929	127,000	77,300	166,000	216,000
H2	9	22,584	54,000	13,900	31,400	8,720
H3	15	3,748	22,900	3,000	19,200	753
H4	21	914	9,730	764	8,810	150

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TABLE 10 MEAN MARSHALL PROPERTIES FOR REJUVENATED ASPHALT MIXES

Binder Composition

	Original Exxon	Oxidized Exxon + REJ G	Oxidized Exxon + REJ C	Oxidized Exxon + REJ D	Original Lion	Oxidized Lion + REJ F	Oxidized Lion + REJ A
Stability	1,495				1,414		
Flow	9				9		
Stability*		1,452	1,381	1,406		1,534	1,576
Flow		11	10	11		10	9
Stability**		1,242	1,283	1,331		1,394	1,405
Flow		10	13	10		9	8

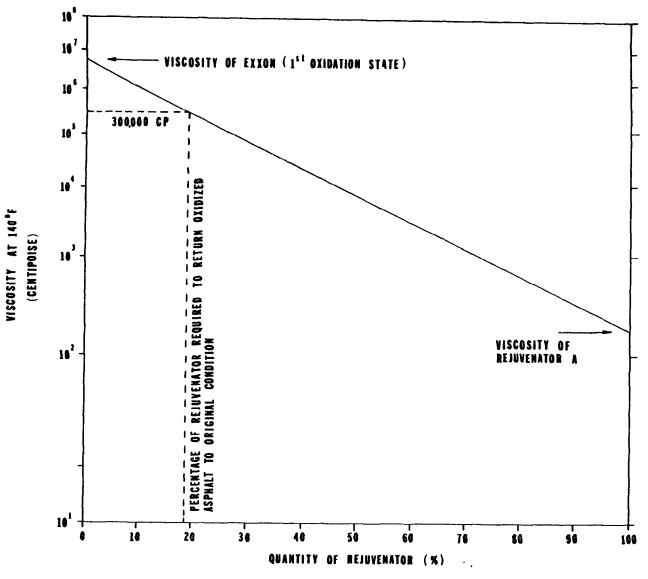
^{*}Mixes comprised of pre-blended oxidized asphalt and rejuvenator.

^{**}Mixes comprised of aggregate coated with oxidized asphalt to which rejuvenator was added.

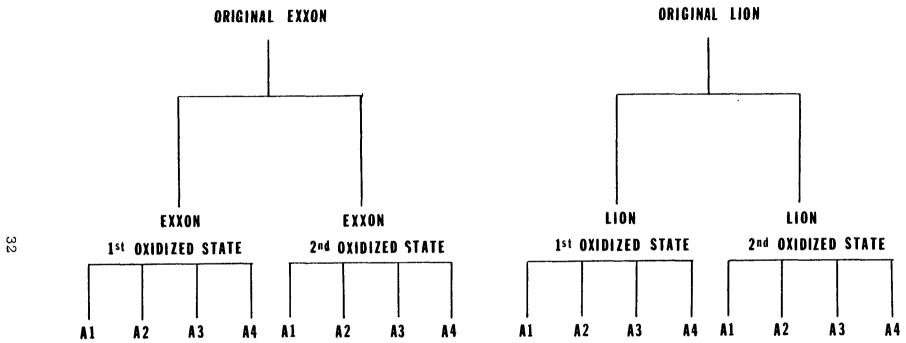
APPENDIX B

LIST OF FIGURES

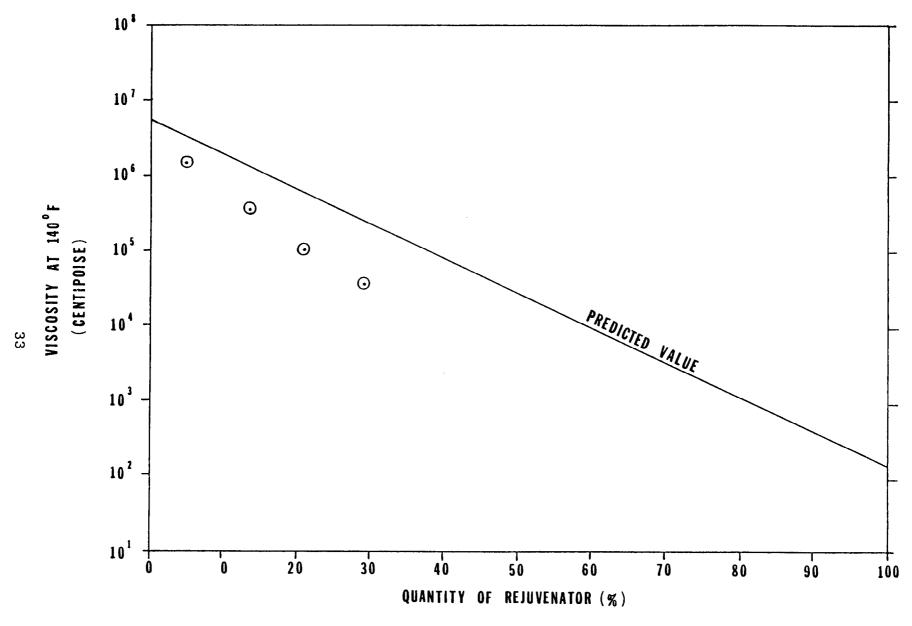
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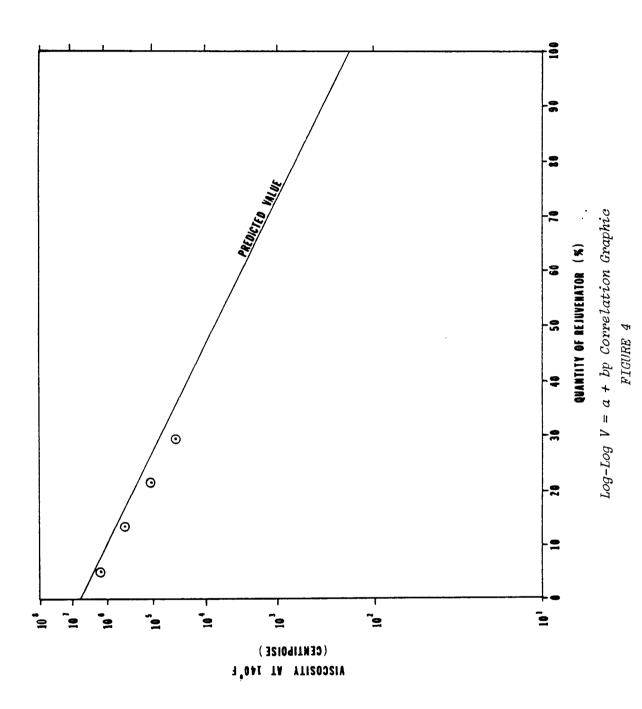
Prediction Equation Graphic Log-Log V = a + bpFIGURE 1

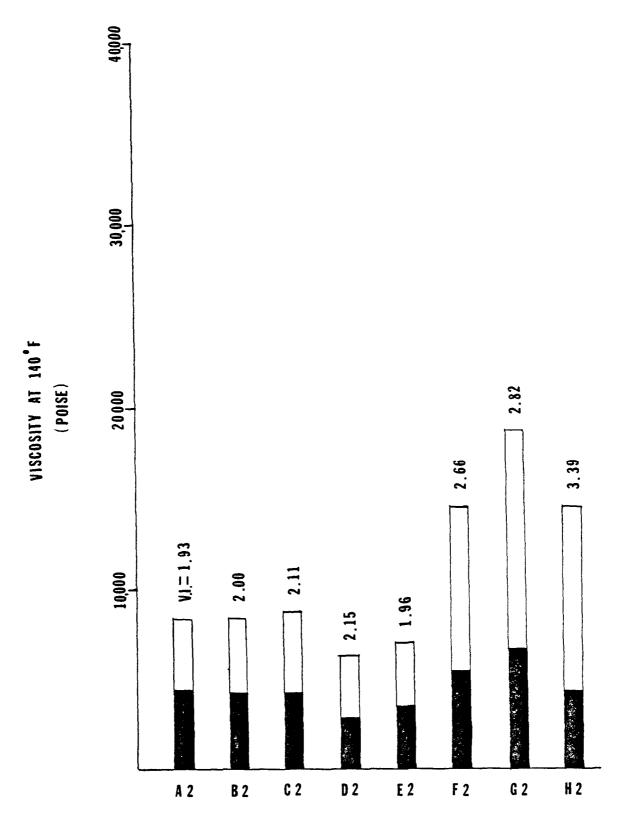


Phase I Testing Matrix FIGURE 2



 $Log\ V = a + bp\ Correlation\ Graphic$ FIGURE 3

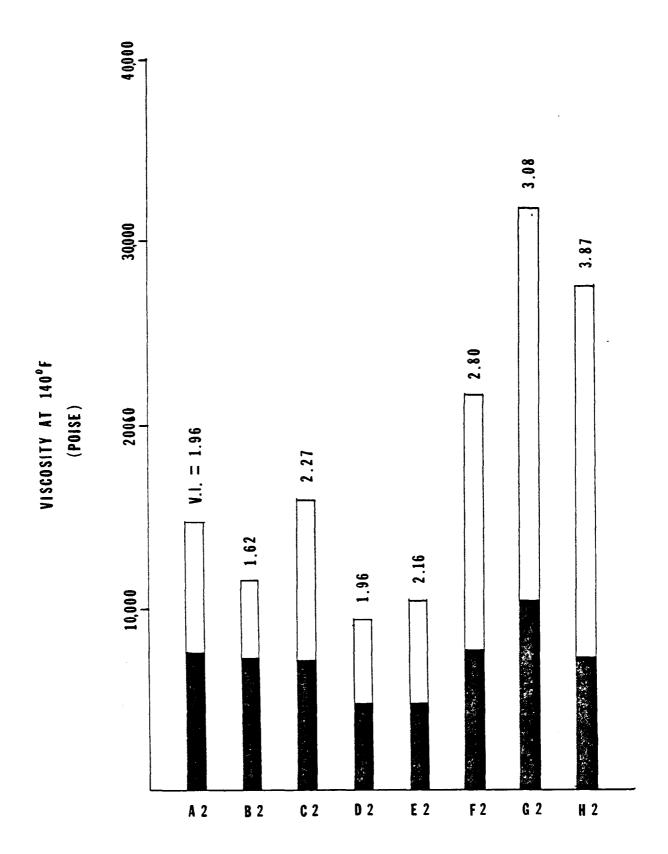




REJUVENATED ASPHALT

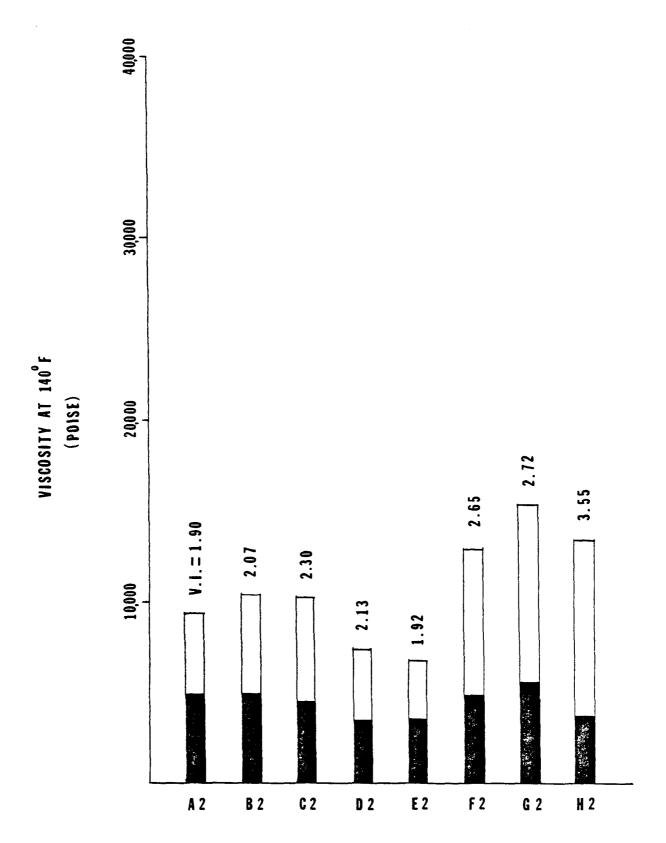
Viscosity Indices for Rejuvenated Exxon -First Oxidation State

FIGURE 5



REJUVENATED ASPHALT

Viscosity Indices for Rejuvenated Exxon -Second Oxidation State FIGURE 6



REJUVENATED ASPHALT

Viscosity Indices for Rejuvenated Lion -First Oxidation State

FIGURE 7

VISCOSITY AT 140°F (POISE)

